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answer. 157.5 g 106.5 g
51 g Yes. Mass of

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reactants equals mass of products. Section 3.2

Assessment page 79 10.

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crushing an aluminum can physical b. recycling used aluminum cans to make new aluminum cans physical c.

aluminum combining with oxygen to form aluminum oxide

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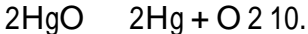
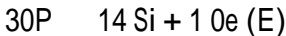
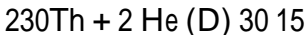
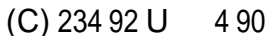
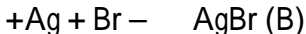
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refer to the following. (A)



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decomposition of a

compound into its ...

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Thinking Critically

Answer the following

questions. Chemistry:

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station has a frequency of

103.7 MHz. (1 MHz 10^6

s⁻¹) What is the

wavelength of the

radiation emitted by the

station? Indicate where

this wavelength falls on

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the electromagnetic
spectrum shown below. c
/ (3.00 10⁸ m/s ...

Answers

Chapter Assessment

mass of oxygen 536.41

214.58 24.36 517.47 g

40% calcium, 12%

carbon, 48% oxygen.

14.58/36.41 540.04%

calcium 4.36/36.41

512.0% carbon

17.47/36.41 547.98%

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51 g Yes. Mass of
reactants equals mass of
products. Section 3.2
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Classify each example as

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Stoichiometry Section

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