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conduction and radiation as well as the use of insulation.

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Temperature. A measure of how hot or cold an object is compared to a reference point. Absolute Zero.

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Work and Heat. Heat -the transfer of thermal energy from one object to another because of a temperature difference
Heat flows from higher temps to lower temps.

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Chapter 16

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488 Chapter 16 What You ' ll Learn You will measure and calculate the energy involved in chemical changes. You will write thermochemical equations and use them to calculate changes in enthalpy. You will explain how changes in

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enthalpy, entropy, and free energy affect the spontaneity of chemical reactions and other processes.

Chapter 16: Energy and Chemical Change

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Thermal Energy and Matter In the 1700 ' s most scientists thought that heat was a fluid called caloric that flowed between objects.

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account; E-mail to a friend; Find other activities; Start over;
Help; A B; heat: the transfer of thermal energy from one
object to another because of a difference in temperature:
temperature: a measure of how hot or cold an object is
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