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MAKHIJANI ~~Class 12th Practical salt~~

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~~detection of chloride and bromide an ions~~

~~Chemistry Practical Viva 3~~ **Setting up**

and Performing a Titration Preparation

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of Pure Sample of Potash Alum - MeitY

OLabs Part 2: Titration of oxalic acid

Ammonium Identification Test Cation Salt

Analysis ~~Titration of oxalic acid with~~

~~KMnO₄ - Class 12 Chemistry C.B.S.E~~

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**Chemistry-Practical-Titration (Mohr's
salt/ KMnO_4) Class 12-(Important**

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Practical) 1/ Introduction to Salt

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~~Molarity and Strength of given KMnO_4~~

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(Mohr's Salt) – Discusses preparation of
250ml of M/20 solution of Mohr's salt
and using that to calculate the molarity
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Add Mohr's salt in small amounts to the

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weighing bottle, so that the pans are balanced. Remove the weighing bottle from the pan. Using a funnel, transfer the Mohr's salt to the volumetric flask. Add about 5ml. of dilute H_2SO_4 to the flask followed by distilled water and dissolve the Mohr's salt.

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CHEMISTRY Salt analysis class 12 1.

SALT ANALYSIS SCHEME OF

ANALYSIS EXPERIMENT

OBSERVATION INFERENCE

PRELIMINARY EXAMINATIONS : 1)

Noted the colour Colourless Absence of

Cu^{2+} , Fe^{2+} , Fe^{3+} , Mn^{2+} , Co^{2+} , Ni^{2+}

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Blue Pre. of Cu^{2+} Brown Pre. of Fe^{3+}
Green Pre. of Ni^{2+} Pale pink Pre. of
 Mn^{2+} Pink Pre. of Co^{2+} 2) Noted the
solubility a) In water b) In dil.HCl a)
Soluble and no ppt with Na_2CO_3 b ...

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colloidal solution (sol) of starch. Exp-2.2 :
To prepare a colloidal solution of gum.

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Exp-2.3 : To prepare colloidal solution (or sol) of egg albumin. Exp-2.4 : To prepare ferric hydroxide, $[\text{Fe}(\text{OH})_3]$ sol. Exp-2.5 : To prepare aluminium hydroxide, [...]

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This is done by conducting a series of tests

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Analysis in a systematic manner and using the observations to confirm the absence or presence of specific cations and anions. Salt analysis is an integral part of the CBSE class 12 chemistry practical examinations and is a topic that several students struggle with.

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~~Salt Analysis: Step by Step Guide with
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is analyzed for the radicals, i.e., cation and

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