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Topics and Subtopics in NCERT Solutions
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NCERT Solutions Haloalkanes and

Haloarenes are chemical compounds

derived from alkanes and arenes

containing one or more halogens. The

NCERT solutions of this chapter broadly
discuss the definitions, classifications,

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~~Solutions~~ examples, and applications and other subtopics under Haloalkanes and Haloarenes.

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The haloalkane is co-heated with an alcohol solution of a strong base, and the

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~~Solutions~~ elimination reaction mainly occurs, and a portion of the hydrogen halide is removed to generate an olefin. When a haloalkane is dehydrohalogenated, hydrogen atoms are mainly removed from adjacent carbon atoms that contain less hydrogen.

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Solutions || Haloalkane ...~~

Class- XII-CBSE-Chemistry Haloalkanes
and Haloarenes Practice more on

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www.embibe.com Solution: (i) Substituent
= CH_3 (Methyl) Secondary prefix = Cl
(Chloro) (Word root) Parent Carbon chain
= 4 (but) Primary Suffix = ane IUPAC
name = Secondary prefix + Primary prefix
+ Word root + primary

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Haloalkanes And Haloarenes Class 12 NCERT Solutions: We know that S N 2

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~~Solutions~~ mechanism involves a transition state wherein both, the incoming nucleophile as well as the leaving group are present around the carbon atom. There are 5 atoms simultaneously bonded to it. Thus, for such a transition state to be possible, there should be minimum steric hindrance.

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halide (haloarene), respectively.

Haloalkanes contain halogen atom(s) attached to the sp^3 hybridised carbon atom of an alkyl group whereas haloarenes contain halogen atom(s) attached to sp^2 hybridised carbon atom(s) of an aryl group. Many halogen containing organic compounds occur in nature and some of these are clinically useful. These classes of compounds

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are also explaine...

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The haloalkanes and haloarenes class 12
notes is a basic note of organic chemistry
because this is the 1st chapter of organic

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Solutions chemistry unit of chemistry class 12. We shall learn a most important topic which is “Replacement of hydrogen atom in alkanes or arenes by 17th group element” in this chapter.

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