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conservation of mass. mass is neither created nor destroyed during ordinary chemical reactions or physical changes $HH+O = HH O$. law of definite proportions. Holt Mcdougal Modern Chemistry Chapter 3 Test Answers

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1. all matter is composed of atoms. 2. atoms of a given element are identical. 3. atoms can't be subdivided, created, or destroyed. 4. atoms of different elements combine in whole-number ratios to form chemical compounds. 5. atoms are combined, separated, or rearranged in chemical reactions. J. J. (Joseph John) Thomson.

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3 Atoms: The Building Blocks of Matter

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