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Li	1.0	900	Be	1.6	801	-27	B	2.0
1086	-122	C	2.6	1402	N	3.0	1314	
-141	+798	O	3.4	1681	-328	F	4.0	
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the Faraday constant F
the Avogadro constant L
the Planck constant h
speed of light in a vacuum c
rest mass of proton, m_p
rest mass of neutron, m_n
rest mass of electron, m_e
electronic charge e

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 $e = 9.65 \times 10^4 \text{ C mol}^{-1} =$
 $6.02 \times 10^{23} \text{ mol}^{-1} = 6.63 \times 10^{-34} \text{ J s} =$
 $3.00 \times 10^8 \text{ m s}^{-1} = 1.67 \times 10^{-27} \text{ kg} =$
 $1.67 \times 10^{-27} \text{ kg} = 9.11 \times 10^{-31} \text{ kg}$
 $=, 1.60 \times 10^{-19} \text{ C}$ molar volume of ...

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