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~~SOLUTIONS MANUAL FOR INTRODUCTION TO THE THERMODYNAMICS OF MATERIALS 6TH EDITION GASKELL~~ Problem 1.1* The plot of $V = V(P, T)$ for a gas is shown in Fig. 1.1.

Determine. the expressions of the two second derivatives of the volume of this plot. (note: the principle curvatures of the surface are proportional to these second derivatives).

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Solution: A volume of 1 L equals 0.001 m³, see Table A.1. From Figure 2.7 the density is in the range of 10 so we get $m \approx 10 \cdot 0.001 \text{ m}^3 = 10 \text{ kg}$ A more accurate value from Table A.4 is 13 580 For the air we see in Figure 2.7 that density is about 1 so we get $m \approx 1 \cdot 0.001 \text{ m}^3 = 0.001 \text{ kg}$ A

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more accurate value from Table A.5 is 1.17 at 100 kPa, 25oC.

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