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CHAPTER 13 REVIEW Ions in Aqueous Solutions and Colligative Properties MIXED REVIEW SHORT ANSWER Answer the following questions in the space provided. 1. Match the four compounds on the right to their descriptions on the left. b an ionic compound that is quite soluble in water (a) HCl c an ionic compound that is not very soluble in water (b) NaNO₃

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CHAPTER 6 REVIEW Chemical Bonding SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. a The notation for sodium chloride, NaCl, stands for one (a) formula unit. (c) crystal. (b) molecule. (d) atom. 2. d In a crystal of an ionic compound, each cation is surrounded by a number of (a) molecules. (c) dipoles. (b ...

6 Chemical Bonding

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Modern Chemistry 105 Chapter Test Name Class Date Chapter Test A, continued Use this figure to answer questions 7 and 8. _____ 7. A solution containing 35 g of Li₂SO₄ dissolved in 100 g of water is heated from 10 ° C to 90 ° C. According to information in the figure, this temperature change would result in a. an additional 5 g of Li₂SO₄ in ...

Assessment Chapter Test A - Ed W. Clark High School

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