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Electrons in Atoms SECTION 1 SHORT
ANSWER Answer the following questions
in the space provided. 1. In what way does
the photoelectric effect support the particle
theory of light? In order for an electron to
be ejected from a metal surface, the
electron must be struck by a single photon
with at least the minimum energy needed
to knock the electron loose. 2. What ...

4 Arrangement of Electrons in Atoms
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molecules of an element found in natural
state. O. oxygen ...

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Class Date Chapter Test B, continued 15.
The energy state of an atom is called its
ground state. 16. The number of waves
that pass a point in one second is called.
17. When an electron drops from a higher-
energy state to a lower-energy state, a(n)
spectrum is produced. 18. The numerical
value of the speed of electromagnetic
radiation in a vacuum is. 19. is ...

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stand for? What value is c ? $c = \text{speed of}$
 light (m/s) 3.00×10^8 $\lambda = \text{wavelength (m)}$
 $v = \text{frequency (Hz or waves/second)}$ List
the 7 types of electromagnetic radiation.
From the equation above, what do all
types of ...

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