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constant motion. Particles are ...

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13.2 The Nature of Liquids The general properties of  
solids reflect the orderly arrangement of their  
particles and the fixed locations of their particles In  
most solids, the atoms, ions, or molecules are packed  
tightly together. When you heat a solid, its particles  
vibrate more rapidly as their kinetic energy increases.

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Chapter 13 states of matter pearson chemistry. in a system at constant vapor pressure, a dynamic equilibrium exists between the vapor and the liquid. The system is in equilibrium because the rate of evaporation of liquid equals the rate of condensation of vapor. Page 5/8

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