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Answer: C. 3. An atom located in the third period of the periodic table would have an electron configuration between $1s^2 2s^2 2p^6 3s^1$ and $1s^2 2s^2 2p^6 3s^2 3p^6$. The highest energy level is 3 as can be seen from the electron configurations. 6.

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Elements and the periodic table. Elements are the purest form of matter and can not be broken down into any other substance by either a physical or chemical change. There are about 114 elements and they are organized on a modern Periodic Table

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of the Elements. The Elements are organized by their atomic numbers from top to bottom and left to right.

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10 points&lots of thumbs up 4 1st person 1. What is the periodic table?

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2. Who developed the first periodic table? 3. How is the modern periodic table organized? 4. What are periods? 5. What are groups or families? 6. What is an element 's atomic number? 8. What is an element 's atomic mass? 9. Why are electrons not included in an element 's atomic

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mass? 10. What is an energy level or shell ...

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